Honors Chemistry Hour\_\_\_\_\_ Name\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_  
Dr. Wexler  
Single Replacement vs. Double Replacement Reactions Worksheet 2  
Date \_\_\_\_\_\_

Complete each word equation, then write them out as balanced reaction equations. Identify each reaction as a single replacement or double replacement reaction.

1. Zinc + silver nitrate🡪

2. Aluminum + hydrogen chloride🡪

3. Magnesium oxalate + ammonium carbonate🡪

4. Calcium + aluminum nitrate🡪

5. Potassium fluoride + lead(II) nitrate🡪

6. Calcium bromide + silver nitrate🡪

7. Ammonium phosphate + barium acetate🡪

8. Sodium chloride + potassium🡪

9. Magnesium nitrate + ammonium chloride🡪

10. Iron (III) chlorate + calcium🡪